Novel Photoreductive Cyclization of 1-Aryl-2-methylpropane-1,3-diones

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Received May 25, 1993®

Irradiation of 2-methyl-1-phenylpropane-1.3-diones la-c in 2-propanol gave trioxatricyclononanes 2a-c, 3a,b, and 4a. Irradiation of 1,3-diaryl-2-methylpropane-1,3-diones 1d and 1e gave trioxatricyclononanes 2d and 2e, dl-pinacols 5d and 5e, and the bicyclic dihemiacetal 6. The dihemiacetal 6 underwent dehydration on heating to give the trioxatricyclononane 3d. Treatment of 2d,e, 5d,e, and 6 with hydrochlolic acid gave dioxabicyclooctadienes 7a and 7b, while that of 2a-c, 3a,b and 4a with hydrochloric acid gave 3-methylenecyclopentenes 8a-c and the epoxycycloheptenone 9.

Photoreduction of aryl ketones by hydrogen-atomdonating substrates has been extensively investigated from both synthetic² and mechanistic³ points of view. A number of α -substituted acetophenones form pinacols upon irradiation in 2-propanol or other suitable hydrogen-donor solvents.² Photoreduction of dicarbonyl compounds has also been investigated. Ethyl 2-benzoylalkanoates, β -keto esters, undergo photoreduction with 2-propanol to give 2,6-dioxabicyclo[3.3.0]octane-3,7-diones via the lactonization of initially formed pinacols.⁴ Irradiation of ethyl acetoacetate in various alcohols gives 4-hydroxy-4,5dihydrofuran-2(3H)-ones via mixed pinacols.⁵ Acetonylacetophenone, a 1,4-diketone, undergoes photoreduction with 2-propanol to give 3,8-diisopropoxy-3,8-dimethyl-1,6diphenyl-2,7-dioxabicyclo[4.4.0]decane via the cyclization of an initially formed pinacol followed by the reaction with 2-propanol.⁶ In relation to our studies on the photoreaction of aryl 1,3-diketones,7 we studied the photoreduction of 1-aryl-2-methylpropane-1,3-diones 1a-e with 2-propanol. We report here that 1,3-diketones 1a-e undergo novel photoreductive cyclization to give 2.6.9trioxatricyclo[3.3.1.0^{3,7}]nonanes 2, 3, and 4 via pinacols and that compounds 2-4 are transformed to 4,8-dioxabicyco[3.3.0]octa-2,6-dienes 7a,b, 3-methylenecyclopentenes 8a-c, and epoxycycloheptenone 9 by treatment with acid.

Results and Discussion

Irradiation of a 2-propanol solution of 2-methyl-1phenylbutane-1,3-dione (1a) with a high-pressure mercury lamp through a Pyrex filter (Scheme I) gave three isomeric trioxatricyclononanes 2a, 3a, and 4a.8 The structures of



these products were elucidated on the basis of analytical and spectral data. The IR spectra showed neither carbonyl nor hydroxy absorptions. The 400-MHz ¹H NMR spectrum of 2a showed peaks due to two CHCH3 groups as two doublets (δ 0.90 and 0.94) and two quartets (δ 2.12 and 2.35), whereas those of 3a and 4a showed a doublet (3a, δ 0.99; **4a**, δ 0.86) and a quartet (**3a**, δ 2.24; **4a**, δ 2.45). The ¹³C NMR spectrum of 2a showed four peaks due to quarternary carbons attached to an oxygen atom at δ 93.8. 94.3, 106.3, and 108.3, whereas those of **3a** and **4a** showed two peaks (3a, δ 93.6 and 106.7; 4a, δ 94.5 and 108.2). These data indicate that 3a and 4a are symmetrical molecules. However, their configurations could not be assigned on the basis of these data. The configuration of 3a was determined by X-ray crystallographic analysis,⁹ which showed that the methyl group on the secondary carbon was trans to the phenyl group. Therefore, the configuration of 4a should be cis with respect to the phenyl and methyl groups.

Irradiation of 2-methyl-1-phenylpentane-1,3-dione (1b) under the same conditions gave trioxatricyclononanes 2b

6737

Abstract published in Advance ACS Abstracts, October 15, 1993. (1) (a) Saitama University. (b) The National Defence Academy. (c) Tokyo Gakugei University.

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and 3b, and that of 2,4-dimethyl-1-phenylpentane-1,3dione (1c) gave 2c. Only one symmetrical trioxatricyclononane and no symmetrical trioxatricyclononanes were obtained from 1b and 1c, respectively, owing probably to steric compression. The configuration of 3b was assigned by comparison of its ¹H NMR spectrum with those of 2a, 3a, and 4a. The ¹H NMR spectrum of 3a showed a doublet due to the C_4 - and C_8 -methyl groups at lower field than those of 2a and 4a. A quartet due to the C_4 - and C_8 methine protons of 3a appeared at higher field than that of 4a and between two quartets of 2a (vide supra). The ¹H NMR spectrum of **3b** showed a doublet (δ 0.94) due to the C_4 - and C_8 -methyl groups at lower field than two doublets (δ 0.88 and 0.89) of **2b** and a quartet (δ 2.28) due to the C₄- and C₈-protons between two quartets (δ 2.15 and 2.35) of 2b.

Irradiation of 2-methyl-1,3-diphenylpropane-1,3-dione (1d) under the same conditions (Scheme II) gave the trioxatricyclononane 2d, the dl-pinacol 5d, and the bicyclic dihemiacetal 6, and that of 2-methyl-1,3-bis(p-methylphenyl)propane-1,3-dione (1e) gave the trioxatricyclononane 2e and the dl-pinacol 5e. The configuration of 6 was elucidated on the basis of dehydration on heating at 200 °C to give the trioxatricyclononane 3d, the configuration of which was assigned by comparison of its ¹H NMR spectrum with those of 2a-d, 3a,b, and 4a (Scheme III). The pinacols 5d,e underwent dehydration to give dioxabicyclooctadienes 7a,b by treatment with hydrochloric acid, probably via the bicyclic dihemiacetal. The treatment of 6 with the acid also gave 7a. The trioxatricyclononanes 2d, e were also converted to 7a, b by the acid.

In contrast to 2d and 2e, when 2a was treated with hydrochloric acid, the 3-methylenecyclopentene 8a and the epoxycycloheptenone 9 were formed in 45 and 9%yields, respectively. The treatment of 3a and 4a with the acid gave the similar results. The compounds 2b and 2c were also converted to 8b and 8c in 37 and 9% yields, respectively. The ¹H NMR spectra of compounds 8 showed two singlets due to exomethylene protons between δ 5.0 and 5.2. The ¹H NMR spectrum of 9 showed an AB quartet due to two methylene protons at δ 2.48 and 2.77. The structure of 9 was finally determined by X-ray crystallographic analysis.⁹ The formation of 8 and 9 can be explained by the initial formation of the dioxabicyclooctadiene 7 which opens the ring to the 3,5-octadiene-2,7dione 10 (Scheme IV). The dione 10 contains a small amount of the isomeric trienone 11. Acid-catalyzed



cyclization of 10 gives 8 and intramolecular [4 + 2] cycloaddition of 11 gives 9.

Photoreduction of aryl 1,3-diketone 1 gives both dl- and meso-pinacols.¹⁰ The formation of trioxatricyclononanes 2-4 can best be rationalized in terms of cyclization of the dl-pinacol to give the bicyclic dihemiacetal and subsequent dehydration, while meso-pinacols or products arising from them could not be isolated. The *dl*-pinacols from 1 have a conformation where the hydroxy groups are located close enough to the carbonyl groups owing to hydrogen bonding so that they cyclize even at room temperature to give bicyclic dihemiacetals which under dehydration to give trioxatricyclononanes 2, 3, and 4 (Scheme V). Three isomers A, B, and C are possible for dl-pinacols from 1. The trioxatricyclononanes 2, 3, and 4 may be formed from A, B, and C, respectively, though these pinacols, except 5d and 5e, could not be isolated. The pinacols 5d and 5e were stable at room temperature. However, when 5d was heated at 180 °C, it underwent quantitative retro-aldol cleavage to give propiophenone and benzil in 2:1 ratio.

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Since two of three possible dl-pinacols from 1d were transformed to trioxatricyclononaes 2d and 3d, the pinacol 5d has configuration C. Cyclization of 5d to the bicyclic dihemiacetal results in significant crowding. Therefore, the pinacol 5d undergoes retro-aldol cleavage in preference to cyclization to the dihemiacetal.

Experimental Section

Melting points were uncorrected. ¹H NMR spectra were recorded at 400 MHz using tetramethylsilane as an internal standard with CDCl₃ as solvent. ¹³C NMR spectra were recorded at 100 MHz with CDCl₃ as solvent. IR spectra were recorded for solutions in CCl₄ unless otherwise stated. A 100-W high-pressure mercury lamp was used as an irradiation source. Compounds 1a and 1d were prepared by the methylation of benzoylacetone and dibenzoylmethane with methyl iodide. Compounds 1b,c,e were prepared by the condensation of the aryl ethyl ketone with the aldehyde according to previously described methods.⁷

General Procedure for Photoreduction. A solution of 1 g of 1 in 120 mL of 2-propanol was irradiated with a high-pressure mercury lamp through a Pyrex filter for 3-10 h. The photoproducts were isolated by silica gel column chromatography using hexane/ethyl acetate (6:1) as eluant. IR spectra of compounds 2, 3, and 4 have neither carbonyl nor hydroxy absorptions.

1,exo-4,5,endo-8-Tetramethyl-3,7-diphenyl-2,6,9-trioxatricyclo[3.3.1.0^{3,7}]nonane (2a): mp 125–126 °C (hexane); ¹H NMR δ 0.90 (3 H, d, J = 7 Hz), 0.94 (3 H, d, J = 7 Hz), 1.54 (3 H, s), 1.60 (3 H, s), 2.12 (1 H, q, J = 7 Hz), 2.35 (1 H, q, J = 7 Hz), 6.8–7.2 (10 H, m); ¹³C NMR δ 9.5 (q), 13.1 (q), 19.6 (q), 20.9 (q), 50.8 (d), 54.9 (d), 93.8 (s), 94.3 (s), 106.3 (s), 108.3 (s), 125.7 (2d), 126.5 (3d), 127.0 (d), 127.3 (2d), 127.7 (2d), 136.2 (s), 138.4 (s). Anal. Calcd for C₂₂H₂₄O₈: C, 78.54; H, 7.19. Found: C, 78.71; H, 7.16.

1,exo-4,5,exo-8-Tetramethyl-3,7-diphenyl-2,6,9-trioxatricyclo[3.3.1.0^{3.7}]nonane (3a): mp 123 °C (methanol); ¹H NMR δ 0.99 (6 H, d, J = 7 Hz), 1.58 (6 H, s), 2.24 (2 H, q, J = 7 Hz), 6.9-7.2 (10 H, m); ¹³C NMR δ 8.4 (2q), 20.7 (2q), 48.6 (2d), 93.6 (2s), 106.7 (2s), 125.4 (4d), 126.7 (2d), 127.5 (4d), 137.8 (2s). Anal. Calcd for C₂₂H₂₄O₃: C, 78.54; H, 7.19. Found: C, 78.43; H, 7.09.

1,endo-4,5,endo-8-Tetramethyl-3,7-diphenyl-2,6,9-trioxatricyclo[3.3.1.0^{4,7}]nonane (4a): mp 154–155 °C (methanol); ¹H NMR δ 0.86 (6 H, d, J = 7 Hz), 1.59 (6 H, s), 2.45 (2 H, q, J = 7 Hz), 6.9–7.2 (10 H, m); ^{13}C NMR δ 13.3 (2q), 19.9 (2q), 55.7 (2d), 94.5 (2s), 108.2 (2s), 126.6 (4d), 126.8 (2d), 127.6 (4d), 136.4 (2s). Anal. Calcd for $C_{22}H_{24}O_3$: C, 78.54; H, 7.19. Found: C, 78.52; H, 7.21.

3,4-Dihydroxy-2,5-dimethyl-1,3,4,6-tetraphenylhexane-1,6-dione (5d): mp178-179 °C (hexane-benzene); IR (KBr) 3570, 3450 br, 1670 cm⁻¹; ¹H NMR δ 0.48 (6 H, d), 4.74 (2 H, s), 4.75 (2 H, q, J = 7 Hz), 6.9-7.9 (20 H, m). Anal. Calcd for C₃₂h₃₀O₄: C, 80.31; H, 6.32. Found: C, 80.17; H, 6.38.

3,7-Dihydroxy-4,8-dimethyl-1,3,5,7-tetraphenyl-2,6-dioxabicyclo[3.3.0]octane (6): mp 191–192 °C (hexane-benzene); IR (KBr) 3300 br cm⁻¹; ¹H NMR δ 1.14 (6H, d, J = 7 Hz), 3.13 (2 H, q, J = 7 Hz), 5.51 (2 H, s), 6.9–7.8 (20 H, m). Anal. Calcd for C₃₂H₃₀O₄: C, 80.31; H, 6.32. Found: C, 80.19; H, 6.34.

Pyrolysis of 6. Compound 6 (40 mg) was placed in a sealed glass tube under reduced pressure (0.4 Torr) and heated at 200 °C for 4 h. The mixture was chromatographed on silica gel (hexane/ethyl acetate 6:1) to give **3d** (28 mg, 72%).

General Procedure for the Treatment of 2-6 with Hydrochloric Acid. A solution of 50 mg of 2-6 in 20 mL of methanol in the presence of 1 drop of concentrated hydrochloric acid was heated under reflux for 3-10 h. The solvent was evaporated and the residue was chromatographed on silica gel (hexane/ethyl acetate (4:1 to 6:1)) to give 7, 8, and/or 9. IR spectra of compounds 7a and 7b have neither carbonyl nor hydroxy absorptions.

2,6-Dimethyl-1,3,5,7-tetraphenyl-4,8-dioxabicyclo[3.3.0]octa-2,6-diene (7a): mp 165-166 °C (hexane); ¹H NMR δ 1.84 (6 H, s), 6.9-8.0 (20 H, m); ¹³C NMR δ 10.1 (2q), 102.8 (2s), 109.3 (2s), 126.8 (2d), 127.0 (4d), 127.4 (4d), 127.6 (4d), 128.3 (4d), 128.5 (2d), 131.7 (2s), 138.7 (2s), 150.7 (2s). Anal. Calcd for C₃₂H₂₆O₂: C, 86.85; H, 5.92. Found: C, 86.57; H, 6.05.

5-Acetyl-4-hydroxy-4,5-dimethyl-3-methylene-1,2-diphenyl-1-cyclopentene (8a): mp 144 °C (hexane); IR 3600, 3450 br, 1710 cm⁻¹; ¹H NMR δ 1.19 (3 H, s), 1.52 (3 H, s), 2.27 (1 H, s), 2.37 (3 H, s), 5.01 (1 H, s), 5.19 (1 H, s), 6.8–6.9 (2 H, m), 7.1–7.2 (5 H, m), 7.2–7.3 (3 H, m); ¹³C NMR δ 15.1 (q), 28.5 (q), 28.6 (q), 69.2 (s), 93.7 (s), 105.4 (t), 127.5 (d), 127.6 (d), 128.0 (2d), 128.3 (2d), 128.8 (2d), 129.7 (2d), 134.8 (2s), 142.1 (s), 145.9 (s), 160.5 (s), 211.0 (s). Anal. Calcd for C₂₂H₂₂O₂: C, 82.99; H, 6.96. Found: C, 83.09; H, 6.94.

3,6-Epoxy-3,4,7-trimethyl-5,6-diphenylcyclohept-4-en-l-one (9): mp 105 °C (hexane); IR 1705 cm⁻¹; ¹H NMR δ 1.03 (3 H, d), 1.56 (3 H, s), 1.63 (3 H, s), 2.48 (1 H) and 2.77 (1 H) (AB system, J = 15 Hz), 2.92 (1 H, q, J = 7 Hz), 6.5–6.6 (2 H, m), 7.1–7.3 (8 H, m) (ArH); ¹³C NMR δ 10.0 (q), 12.7 (q), 21.7 (q), 47.0 (t), 49.8 (d), 84.5 (s), 90.1 (s), 126.0 (2d), 127.2 (d), 127.4 (d), 127.8 (4d), 129.6 (2d), 132.9 (s), 137.9 (s), 139.2 (s), 143.2 (s), 211.1 (s). Anal. Calcd for C₂₂H₂₂O₂: C, 82.99; H, 6.96. Found: C, 82.92; H, 6.95.

Supplementary Material Available: Compound characterization data for 2b-e, 3b, 3d, 5e, 7b, 8b and 8c and ORTEP diagrams of 3a and 9 (5 pages). This material is contained in libraries on microfiche, immediately follows this article in the microfilm version of the journal, and can be ordered from the ACS; see any current masthead page for ordering information.